A TANDEM 1.3-DIPOLAR CYCLOADDITION/ELECTROPHILIC CYCLIZATION SEQUENCE: γ ₅-UNSATURATED ISOXAZOLINES AS PRECURSORS TO RING-FUSED **ETHERS**

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Abstract: A tandem 1,3-dipolar cycloaddition/electrophilic cyclization sequence affords cis-fused *octahydrobenzo- and hexahydro-2H-cyclopenta[bjfkrans via the intermediaq of x&unsaturated isoxazolines.*

We recently described a new strategy for the construction of tetrahydrofurans by bis-addition of an oxygen nucleophile across a 1,5-diene moiety.² That study, in which functionalized 2,5-disubstituted tetrahydrofurans were obtained from 1,5-hexadiene by a tandem 1,3-dipolar cycloaddition/electrophilic cyclization sequence, has opened up a variety of promising avenues for the synthesis of other cyclic ethers. Herein, we report the application of this isoxazoline-based strategy to the synthesis of cis-fused octahydrobenzo- and hexahydro-2Hcyclopenta[b]furans; substructures of considerable importance in natural products chemistry.3

In contrast to a simple starting substrate like 1,5-hexadiene which delivers 2-(cyanomethyl)-5-(iodomethyl)tetrahydrofuran, employing an unsymmetrical 1,5-diene as the starting material for this two-step sequence affords the potential to construct more elaborate cyclic ethers. However, this extension introduces two new issues: namely, the questions of regio- and chemoselectivity in the 1,3-dipolar cycloaddition step. It is generally found that the rate of nitrile oxide•alkene 1,3-dipolar cycloaddition increases as olefin substitution (i.e., steric hinderance) decreases. Indeed nitrile oxides generally react selectively with a terminal double bond in the presence of an internal double bond and cyclic alkenes, such as cyclohexene and cyclopentene, show higher dipolarophilicity than the corresponding 1,2-disubstituted acyclic olefins.⁴ Thus exploiting this 1,3-dipolar cycloaddition/electrophilic cylization sequence in the construction of more elaborate cyclic ethers necessitates employing starting 1,5-dienes in which the two olefins differ markedly in dipolarophilicity.

As illustrated in **Scheme I,** 1,5-dienes **la** and **lb** both require olefin selectivity in the 1,3-dipolar cycloaddition step. Fortunately, triphenylacetonitrile oxide⁵ is a highly selective 1,3-dipolar reagent⁶ and, as hoped, undergoes cycloaddition exclusively on the terminal olefin with complete regioselectivity. An added consequence of this olefin selectivity is that, unlike symmetrical dienes where a large excess of 1,3-dipolarophile must be employed to avoid the formation of bis-cycloadducts, these unsymmetrical dienes can be used in stoichiometric amount, giving isoxazoline 2 in excellent yield (1a->2a in 80% and 1b->2b in 85%). Subsequent intramolecular cyclization with iodine in CH₂Cl₂ under gentle reflux gave the desired oxabicyclic tetrahydrofuran derivatives $(3 + 4)$ as a mixture of diastereomers.

Iodocyclization of 2 proceeds regiospecifically producing either the $[4.3.0]$ $(2a-3a + 4a)$ in 82% yield) or the [3.3.0] **(2b+3b + 4b in 59%** yield) oxabicycle with exclusive *cis* ring-fusion (complete internal asymmetric induction). These ring-fusion stereochemical assignments were made on the basis of literature ¹H-NMR data for the octahydrobenzofuran ring system and by analogy in the hexahydro- $2H$ -cyclopenta[b]furan ring system.⁷ Thus, H-7a and H-2 in **6a** and H-6a and H-2 in 6b are deshielded relative to those protons in **5a** and 5b, respectively (see **Table I). In** addition, 2-(2-cyclopenten-1-yl)ethanol is reported to undergo electrophilic cyclization with complete cis ring-fusion selectivity.⁸ Endo/exo stereochemical assignments in 3 and 4 follow from the corresponding assignments in 5 and 6 , in turn prepared by Bu₃SnH/AIBN deiodination. These C-2 stereochemical assignments were further verified by 2-D NOESY correlation techniques. For example, while 5a and **6a** both show an H-2 \Leftrightarrow CH₂C=N NOE cross peak, only **6a** shows an H-7a \Leftrightarrow CH₂C=N NOE cross peak. The implications of this observations, spatial proximity between H-7a and the CH₂C=N moiety in 6a but not 5a, **are in accord with these** chemical shift based stereochemical assignments.

<u>proton</u>	<u>5a</u>	<u>6ª</u>	$\overset{\text{5b}}{=}$	亝
$H-2$	4.18δ	4.42δ	4.45δ	4.67δ
$H-7a$	3.90δ	4.13 δ	-1	--
H-6a	$- -$	\bullet	3.92δ	4.31 δ

Table I. Partial **300 MHz** IH-NMR Data for **5** and 6.

ln **7a, the** next substrate investigated, triphenylacetonitrile oxide was not capable of clearly differentiating the mono- and disubstituted olefins and the anticipated isoxazoline product (8) was contaminated with \approx 30% of spiro isoxazoline 9. To circumvent this selectivity problem, the offending disubstituted olefin was not introduced until after the 1.3-dipolar cycloaddition step. This was accomplished via a modified Peterson olefination⁹ 'protection sequence' starting from 2allylcyclohexanone **(7b)** as outlined in **Scheme** II. The 1,3-dipolar cycloaddition reaction of 10 delivers isoxaxoline 11 which, upon exposure to concentrated sulfuric acid/THF at room temperature, leads to 8 in 57% overall yield from 7b. As anticipated, subsequent treatment with iodine in CH₂Cl₂ under gentle reflux provided the desired cis-fused tetrahydrofuran 12 in moderate yield (45%).

As a final demonstration of the synthetic versatility of this tandem 1,3-dipolar cycloaddition/electrophilic cyclixation sequence, iodoethyl tetrahydrofurans 15a.b were easily prepared starting from dienes **13a,b.'o In** this case, cycloaddition proceeds exclusively and in **quantitative yield on the terminal olefin, not on the more** hindered trisubstituted exocyclic double bond, giving isoxazoline 14 as a mixture of diastereomers (13a->14a in 84% yield; 13b→14b in 71% yield). Subsequent iodocyclization with iodine in CH₂Cl₂ under gentle reflux **gives 15. None** of the alternate tetrahydropyran products were detected in the crude reaction mixture.

Scheme III

In conclusion, the synthetic potential of this tandem 1,3-dipolar cycloaddition/electrophilic cyclization sequence in the construction of highly functionalized cyclic ethers is evident. The methodology reported here complements existing methods for the preparation of oxabicyclic ring systems¹¹ and extends the utility of isoxazoline-based chemistry.

Experimental Section

General Experimental. Infrared spectra were determined on a IBM FTR-32 with IBM 9000 data system. NMB Spectra were **determined on a Vanan EM390 spectrometer** (lH at 90 MHz), or **a General Electric** QE-300 spectrometer (lH at 300 MHz and 13C at **75** MHz). Mass spectra were determmed on a Dupnnt 21492B mass spectrometer. Melting points were determined on a Thomas Hoover Uni-Melt melting point apparatus and are uncorrected. MPLC refers to column chromatography done at 10-50 psi through EM Lobar^m columns packed with LiChroprep S160 (40-63 µm) or prepared columns packed with Florisil^m (60-100 mesh) with hexane/EtOAc eluent and monitored by refractive index detection. Chromatron refers to preparative, centrifugally accelerated, radial, **thin-layer chromatography with sihca gel 60 as stationary phase. Analytical thin layer chromatrogmphy (TLC) was performed with** Kodak 100 micron thick silica gel plates. Capillary gas chromatography (GC) was performed on a Hewlett-Packard 5890A gas chromatograph using a DB-1701 column (30 m X .259 mm; film thickness = 0.25 mm): initial temperature = 90° C; initial time = 1 min; rate = 2° C/min; gas pressures (psi: He = 60; N₂ = 32; air = 34; H₂ = 20).

3-Triphenplmethyl-5-[(2-cyclohexen-l-yl)methyl]-2-isoxazoline (2a): General Procedure A. An anhydrous benzene (25 mL) solution of triphenylmethylacetonitrile oxide (970 mg, 3.39 mmol) and 3-allykyclohexene (440 mg, 3.60 mmol) in a scaled tube was heated at 85°C for 48 h. After cooling to room temperature, the crude reaction mixture was concentrated under reduced pressure and purified by MPLC (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) giving 2a (1.1 g, 80%) as a white solid [mp 125-127°C (benzene); R_f = 0.83 (20:80 EtOAc:hexane); diastereomeric mixture: FT-IR (KBr) 3202, 2927, 2295, 1597 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.21-2.37 (m, 9 H), 2.44 (dd, J = 9, 14 Hz, 1 H, N=C-CHH), 2.92 (dd, J = 9, 14 Hz, 1 H, N=C-CHH), 4.70 (m, lH, HCO), 5.51 (m. 1H. CH=CH). 5.81 (m. 1 H. CH=CH). 7.51 (m, 15H. Ar-H). Anal. Cakd for Cg9H2eNO: C, 85.46; H. 7.18; N. 3.44. Found: C. 85.18; H, 7.11; N, 3.291.

3-Triphenylmethyl-5-[(2-cyclopenten-l-yl)methyl]-2-isoxazoline (2b). As in Prucedure **A.** triphenylmethylacetonitrile oxide (894 mg, 3.13 mmol), 3-allylcyclopentene (388 mg, 3.59 mmol), and MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave 2b (1.05 g, 85%) as a white solid [mp 108-110°C (benzene); Rf = 0.80 (20:80 EtOAc:hexane); diastereomeric mixture: FT-IR (KBr) 3059, 2942, 2853, 1597 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.48 (m, 2 H), 1.85 (m, 1 H), 2.18 (m. 1 II). 2.42 (m. 2 H). 2.60 (dd, J = 9.15 Hz. 1 H, N=C-(XI-I), 2.71 (m. 1 H), 2.95 (dd, J = 9,15 Hz, 1 H. N=C_cHED, 4.68 (m, 1 H, HCO), 5.78 (m, 2 H), 7.51 (m, 15H, Ar-H); HRMS calcd for C₂₈H₂₇NO 393.2093, found 393.2092].

 $(1S^*.SR^*.SR^*.RR^*)$ - and $(1S^*.SR^*.RR^*.8S^*)$ -5-Iodo-8-cyanomethyl-7-oxabicyclo[4.3.0]-nonane (3a) and (4a): **General Procedure B.** Solid iodine (1.24 g, 4.89 mmol) was added in one portion to a methylene chloride (40 mL) **solution** of isoxaxoline 2a (997 mg, 2.46 mmol) and the resulting mixture heated at reflux under nitrogen for 36 h at which time the starting isoxazoline was no longer detectable by thin layer chromatography. Ether (30mL) and a saturated aqueous Na₂S₂O₅ solution (10 mL) were added to the cooled reaction mixture, the layers were separated and the aqueous phase was extracted with ether (3 x 25 mL). The combined organics were washed with brine (10 mL), dried (MgSO4), filtered, and concentrated under reduced pressure. Purification of the resulting oil by MPLC (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave, in order of elution, 3a (185 mg, 26%) [R_f = 0.67 (25:75 EtOAc:hexanes); FT-IR (neat) 2878, 2251 (CN), 1061, 1096 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.41-2.31 (m, 9 H), 2.57 (m, 1 H), 2.71 (m, 2 H, CH₂CN), 4.29 (m, 2 H, 2 HCO); ¹³C (CDCl₃) δ 22.6, 25.9, 27.2, 32.6, 33.3, 36.9 (HCI), 37.1, 74.2 (HCO), 84.8 (HCO), 119 (CN); HRMS calcd for $C_{10}H_{14}INO$ 291.0122, found 291.0119] and 4a (369 mg, 56%) [R_f = 0.50 25:75 EtOAc:hexane); FT-IR (neat) 2878, 2251 (CN), 1061, 1097 cm-1; ¹H NMR (300 MHz, CDCl₃) δ 1.25-2.30 (m. 9 H), 2.60 (m, 3 H, CH₂CN/**H**CHCO), 4.41 (m, 2 H, 2 HCO); ¹³C (CDCI₃) δ 23.8, 25.7, 27.3, 34.3, 35.5, 36.3, 39.3 (HCI), 74.9 (HCO), 85.8 (HCO), 119 (CN); HRMS calcd for $C_{10}H_{14}$ INO 291.0122, found 291.0120. Anal. Calcd for $C_{10}H_{14}$ INO: C, 41.26; H, 4.85; N, 4.81. Found: C, 41.75; H, 5.12; N, 4.951.

 $(1R^*,3R^*,5S^*,8R^*)$ - and $(1R^*,3S^*,5S^*,8R^*)$ -8-Iodo-3-cyanomethyl-2-oxabicyclo[3.3.0]-octane (3b) and **(4b). As in Procedure A, iodine** (1.07 g. 4.22 mmol) and 2b (830 mg, 2.11 mmol) followed by MPLC purification (25:75 EtOAc:hexanes, 2.5 mL/min, RI detector) gave an inseparable 1:1.7 mixture of 3h and 4b (340 mg, 59%) as a yellow oil [diastereomeric mixture: FT-IR (neat) 2871, 2251 (CN), 1072 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.87-2.82 (m, 8H), 2.86-3.19 (m, 1H), 3.80-4.29 (m, 1H), 4.29-4.99 (m, 2H); ¹³C-NMR (CDCl₃) δ 24.9, 32.1, 33.8, 37.7, 40.3, 42.4 (HCl), 75.3/76.7 (HCO), 94.6/95 (HCO), 118.1/118.3 (CN); HRMS calcd for C9H₁₂INO 276.9965, found 276.9956].

(lS*,6R*,8R*)-8-Cyanomethyl-7-oxabicyclo[4.3.0]nonane (5a): General Procedure C. A benzene **(2** mL) solution of **3a** (40 mg, 0.14 mmol), Bu3SnH (60 mg, 0.21 mmol), and AIBN (2.2 mg, 0.014 mmol) was refiuxed for 12 h under nitrogen. After cooling and concentration of the mixture under reduced pressure, ether (1OmL) and saturated aqueous KF (2mL) were added and the mixture allowed to star for 1 h at room temperature. The layers were separated, the aqueous phase was extracted with ether (3 x 25 mL), and the combined organics were washed with brine (10 mL), dried (MgSO4), filtered, and concentrated under reduced pressure. Purification of the resulting oil by MPLC (20:80 EtOAc:hexanes. 2.5 mL/min, RI detector) gave Sa (19 **mg, 85%) as a** yellow oil [Rf = 0.45 (25:75 EtOAc:hexanes); FT-IR (neat) 2932, 2858, 2249 (CN), 1111 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.2-2.0 (m. 9 H), 2.21 (m. 2H), 2.63 (dd, J = 5. 13 Hz, 1 H, CHHCN), 2.69 (dd, J = 5. 13 Hz, I H. CHHCN), 3.90 (m, 1 H. HCO). 4.18 (m, 1 H, HCO); HRMS calcd for C₁₀H₁₅NO 165.1154, found 165.1156].

(1S*,6R*,8S*)-8-Cyanomethyi-7-oxabicyclo[4.3.O]nonane (6a). As in Procedure C, 4a (61 mg, 0.21 mmol), Bu3SnH (91 mg, 0.32 mmol), and AIBN (3.3 mg, 0.021 mmol) followed by MPLC purification (20:80 EtOAc:hexanes, 2.5 mL/min, **RI detector) gave 6a (28 mg, 80%) as a yellow oil [FT-IR (neat) 2932, 2858, 2249 (CN), 1111 cm⁻¹; ¹H-NMR (300 MHz, CDCl₃)** δ 1.21-2.25 (m, 11 H), 2.48 (dd, J = 4.5, 15 Hz, 1 H, CHHCN), 2.60 (dd, J = 4, 15 Hz, 1 H, CH<u>H</u>CN), 4.13 (m, 1 H, HCO), 4.42 (m, 1 H, HCO); HRMS calcd for C₁₀H₁₅NO 165.1154, found 165.1153].

 $(1R^*,3R^*,5S^*)$ - and $(1R^*,3S^*,5S^*)$ -3-Cyanomethyl-2-oxabicyclo[3.3.0]octane (5b) and (6b). As in **Procedure C. a 3b/4b mixture (21.8 mg, 0.079 mmol). Bu3SnH (34.0 mg. 0.118 mmol). and AlBN (1.3 mg. 0.008 mmol)** followed by MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave, in order of elution, 5b (4.1 mg, 37%) [Rf = 0.45 (20:80 EtOAc:hexanes); FT-IR (neat) 2961, 2870, 2251 (CN), 1049 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.2-2.0 (m, 7 H), **2.35 (m, 1 H). 2.62 (m, 2 H, CH2CN). 2.68 (m. 1 H), 3.92 (d, J = 4.4 Hz. 1 H HCO). 4.45 (1. J = 6.5 Hz, 1 H. HCO); HRMS calcd** for C₉H₁₃NO 151.0997, found 151.0997] and 6b (3.1 mg, 44%) $[R_f = 0.3 \ (20.80 \text{ EtOAc:hexanes})$; FT-IR (neat) 2961, 2871, 2251 **(CN). 1049 cm-l; IH-NMR (300 MHz, cDcl3) 8 U-2.0 (m. 8 H), 2.58 (m, 2 H, CH2CN). 2.81 (m. 1 H). 4.31 (q. J = 6.4 Hz, 1** H. HCO), 4.67 (bt, J = 4.5, 1 H, HCO); HRMS calcd for C₉H₁₃NO 151.0997, found 151.0994] as yellow oils.

Trimetbyl[(l-hydroxy-2-allylcyclohexyl)methyl]silane (10) IO. ZAllylcyclohexanone (4.0 g, 28.9 mmol) in ether (20 mL) was added dropwise via a syringe pump to a 0°C solution of trimethylsilylmethylmagnesium bromide (72.3 mmol) in anhydrous ether (30 mL). When the addition was complete, the mixture was heated at 40°C for 48 h, then cooled to 0°C and quenched with 5% sodium hydroxide. After stirring the mixture at room temperature for 1 h, the layers were separated, the aqueous phase was **exmted with ether (3 x 30 ml), and the combined organic extracts were washed with brine (1 x 20 mL). dried (Na2SO4). filtered, and** concentrated under reduced pressure. Purification of the resulting oil by Kugelröhr distillation (230^oC) gave 10 (6.5 g, 99%) as a pale yellow oil [FT-IR (neat) 3200-3600 (OH), 2934, 1248, 862 cm⁻¹; ¹H-NMR (300 MHz, CDCl₃) 8 0.18 (s, 9 H, 3 CH₃), 1.0-1.7 (m, 12 H), 1.9 (m, 1 H), 2.4 (m, 1 H, OH), 5.0 (m, 2 H, CH=CH₂), 5.8 (m, 1 H, CH=CH₂)].

3-Tripheaylmethyl-S-[((2-hydroxyl-2-trimethylsilyoxymethyl)-cyclohex-l-yl)methyl]-2-isoxazoline (11). As in Proeedure A, triphenylmethylacetonitrile oxide (1.34 g. 4.69 mmol) and 10 (1.06 g, 4.69 mmol) followed by MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave 11 (1.9 g, 80%) as a colorless oil [Rf = 0.75 (25:75 EtOAc:hexanes): diastereomeric mixture: FT-IR (neat) 3850, 3518, 3059, 2855, 1034 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 0.35 (s, 9H), 0.85-2.01 (m, 14H), 2.45 (m. IH, N=C-CHH). 2.93 (s, lH, N=C-CHH.), 4.68 (m. 1H. HCO), 7.54 (m. 15H); HRMS calcd for C33H41NO2Si 511.2906. found 511.29051.

3-Triphenylmethyl-5-[((2-methylene)-cyclohex-l-yl)methyl]-2-isoxazoline (8). Concentrated sulfuric acid (0.92 g) was added to a tetrahydrofuran (20 mL) solution of isoxazoline 11 (1.62 g, 3.3 mmol) and the solution was allowed to stir at room **temperature for 48 h. Ether (30mL) and water (2OmL) were then added, the layers were separated. and the aqueous phase was extracted** with ether (3 x 30 mL). The combined organic extracts were washed with 10% aqueous NaHCO₃ (10 mL) and brine (10 mL), dried **(K2CO3). filtered, and concentrated under reduced pressure. Purification of the resulting oil by MPLC (4% EtOAc:hexanes, 2.5 mUrnm, RI detector) gave 8 (1.0 g. 72%) as a colorless oil** [Rf = **0.64 (10~90 EtOkhexanes); m-lR (neat) 3061,2928,1643,1493, 1084 cm-l; lH-NMR (300 MHz. CDCl3) 8 1.08-2.30 (m. llH), 2.41 (dd. J = 4.5, 13 Hz, 1 H, N=C-CEfH), 2.90 (m. 1 H, N=C-**CHH), 4.55 (bd, J = 4.5 Hz, 1 H, C=CHH), 4.60 (bd, J = 4.5 Hz, 1 H, C=CHH), 7.42 (m, 15H); HRMS calcd for C30H31NO **421.2406, found 421.24131.**

 $(1S^*, 6R^*, 8R^*)$ - and $(1S^*, 6R^*, 8S^*)$ -8-Cyanomethyl-6-iodomethyl-7-oxabicyclo[4.3.0]-nonane (12). As in Procedure A, iodine (1.15 g, 4.3 mmol) and 8 (926 mg, 2.19 mmol) followed by MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave 12 (1:3 mixture as a yellow oil, 288 mg, 45%) as an inseparable mixture of diastereomers [R_f = 0.67 (25:75 EtOAc:hexane); diastereomeric mixture: FT-IR (neat) 2965, 2249 cm⁻¹; ¹H-NMR (300 MHz, CDCl₃) δ 1.08-2.31 **(complex, 13 H), 2.63 (d, J = 6.8, 2 H, CH2CN). 4.2-4.4 (m, 1 H. HCO); lsC (CDCls) 8 20.3/20.7. 22.2fZ2.4. 24.1, 24.6, 24.8,** 24.9, 25.1/25.6, 33.3/34.3, 35.2/35.4, 41.4/42.9, 70.3/71.3 (HCO), 80.2/81.1, 119 (CN); HRMS calcd for C₁₁H₁₆NO (-I) Fab+ **178.1232. found 178.12301.**

3-Triphenylmethyl-5-[((2-ethylidene)-cyclohex-l-yl)methyl]-2-isoxazoline (14a). As in Procedure A, triphenylmethylacetonitrile oxide (1.0 g, 3.5 mmol) and 13a (540 mg, 3.6 mmol) followed by MPLC purification (5:95 EtOAc:hexanes, 2.5 mL/min, RI detector) gave 14a (colorless oil, 1.3 g, 84%) as an inseparable mixture of diastereomers [R_f = 0.64 $(10:90 \text{ EtOAc:hexane})$; FT-IR (neat) 3059, 2855, 1493, 1034 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.05-3.12 (m, 14 H), 2.12 (m, 1 H. N=C-CHH), 2.77 (dd. J = 7, 13 Hz, 1 H, N=C-CHH), 4.45 (m, 1 H, HCO), 5.01-5.23 (m, 1 H, CH₃-CH=), 7.43 (m, 15 H, Ar-**H); HRMS calcd for C31H33NO 435.2562, found 435.2560].**

3-Triphenylmethyl-5-[((2-ethylene)-cyclopent-1-yl)methyl]-2-isoxazoline (14b). As in Procedure A, triphenylmethylacetonitrile oxide (134 mg, 0.47 mmol) and 13b (71 mg, 0.58 mmol) followed by MPLC purification (2:98 EtOAc:hexanes, 2.5 mL/min, RI detector) gave 14b (colorless oil, 120 mg, 71%) as an inseparable mixture of diastereomers [Rf = 0.54 (10:90 EtOAc:hexane); FT-IR (neat) 3032, 2862, 1597, 1493 cm⁻¹; ¹H-NMR (300 MHz, CDCl3) δ 1.46-2.2 (m, 12 H), 2.35 (m, 1 H), 2.93 (m, 1 H), 4.65 (m, 1 H, HCO), 5.20-5.41 (m, 1 H, CH3-CH=), 7.50 (m, 15 H, Ar-H); HRMS calcd for C30H31NO 421.2406, found 421.2406].

 $(1S^*, 6R^*, 8R^*)$ - and $(1S^*, 6R^*, 8S^*)$ -8-Cyanomethyl-6- $(2$ -iodoethyl)-7-oxabicyclo[4.3.0]-nonane (15a). As in Procedure A, iodine (0.99 g, 3.9 mmol) and 14a (0.85 g, 1.82 mmol) followed by MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave, in order of elution, endo-15a (71 mg, 12%) [R $_f = 0.66$ (25:75 EtOAc:hexane); FT-IR (neat) 2930, 2860, 2251 (CN) 1415, 1091 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.95 (t, J = 7 Hz, 3 H, CH₃), 1.20-2.23 (m, 10 H), 2.18 (m, 2 H), 2.63 (d, J = 4 Hz, 2 H, CH₂CN), 4.18 (m, 1 H, HCO); ¹³C (CDCl₃) δ 8.8 (CH₃), 22.0, 23.8, 26.5, 26.8, 31.1, 32.5, 36.0, 40.6, 72.1 (HCO), 88.1, 119 (CN); HRMS calcd for C12H18INO (-I) 192.1388, found 192.1389] and exo-15a (129 mg, 22%) [Rf = 0.50 (25:75 EtOAc:hexane); FT-IR (neat) 2930, 2862, 2251 (CN), 1414, 1092 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.95 (t, J = 7 Hz, 3 H, CH₃), 1.2-2.0 (m, 10 H), 2.21 (m, 2 H), 2.56 (d, J = 5 Hz, 2 H, CH₂CN), 4.3 (m, 1 H, HCO); ¹³C (CDCl₃) δ 6.9 (CH₃), 20.6, 21.9, 24.0, 25.5, 28.9, 32.9, 35.2, 40.1, 71 (HCO), 83, 118 (CN); HRMS calcd for C₁₂H₁₈INO (-I) 192.1388, found 192.1382].

 $(1R^*, 3R^*, 5S^*)$ - and $(1R^*, 3S^*, 5S^*)$ -3-Cyanomethyl-1-(2-iodoethyl)-2-oxabicyclo[3.3.0]-octane (15b). As in Procedure A, iodine (120 mg, 0.47 mmol) and 14b (100 mg, 0.24 mmol) followed by MPLC purification (10:90 EtOAc:hexanes, 2.5 mL/min, RI detector) gave, in order of elution, endo-15b (21 mg, 28%) [Rf = 0.66 (25:75 EtOAc:hexane); FT-IR (neat) 2868, 2253 (CN), 1464, 1095 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.0 (t, J = 7 Hz, 3 H, CH₃), 1.22-1.80 (m, 9 H), 2.39 (m, 1 H), 2.61 (m, 2 H, CH₂CN), 4.0 (m, 1 H, HCO); ¹³C (CDCl₃) δ 10.0 (CH₃), 26.6, 31.8, 33.5, 34.3, 39.8, 40.3, 47.5 (HCI), 73.1 (HCO), 96.5, 117.3 (CN); HRMS calcd for C11H16INO (+H) FAB+ 306.0357, found 306.0358] and exo-15b (23 mg, 31%) [R_f = 0.50 (25:75 EtOAc:hexane); FT-IR (neat) 2868, 2253 (CN), 1464, 1095 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.93 (t, J = 7 Hz, 3 H, CH₃), 1.3-2.0 (m, 9 H), 2.39 (m, 1 H), 2.60 (m, 2 H, CH₂CN), 4.2 (m, 1 H, HCO); ¹³C (CDCl₃) δ 9.1 (CH₃), 23.9, 25.6, 32.5, 33.3, 38.8, 39.3, 46.5 (HCI), 73.7 (HCO), 96.6, 117.3 (CN); HRMS calcd for C₁₁H₁₆INO (+H) FAB+ 306.0357, found 306.0358].

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References

- Sloan Fondation Fellow, 1987-1989. $\mathbf{1}$
- Kurth, M.J.; Rodriguez, M.J. J. Am. Chem. Soc. 1987, 109, 7577. $\mathbf{2}$
- 3 (a) Nicolaou, K.C.; Duggan, M.E.; Hwang, C.-K.; J. Am. Chem. Soc. 1986, 108, 2468. (b) Nicolaou, K.C.; Duggan, M.E.; Hwang, C.-K.; Somers, D.K. J. Chem. Soc., Chem. Commun. 1985, 1359. (c). Shimizu, Y.; Chou, H.-N.; Bando, H.; VanDuyne, G.; Clardy, J.C. J. Am. Chem. Soc. 1986, 108, 514.
- Zinner, G.; Gunther, H. Chem. Ber. 1965, 98, 1353. 4
- 5 Wieland, H.; Rodenfeld, B. Justus Liebigs Ann. Chem. 1930, 484, 236.
- Grundmann, C.; Grunanger, P. The Nitrile Oxides: Versatile Tools of Theoretical and Preparative Chemistry; Springer-Verlag, 6 New York, 1971.
- $\overline{7}$ Mihailovic, M.L.J.; Jeremic, D.; Milosavljevic, S.; Gojkovic, S.; Andrejevic, V. Vestn. Slov. Kem. Drus. 1986, 295.
- 8 (a) Murata, S.; Suzuki, T. Tetrahedron Lett. 1987, 28, 4415. (b) Clive, D.L.J.; Chittattu, G.; Wong, C.K. Can. J. Chem. 1977, 55, 3894.
- 9 Peterson, D.J. J. Org. Chem. 1968, 33, 780.
- 10 Hauser, C.R. and Hance, C.R. J. Am. Chem. Soc. 1952, 74, 5091.
- 11 (a) Hoye, T.R.; Caruso, A.J. J. Org. Chem. 1981, 46, 1198. (b) Smith, A.B., III; Schow, S.R.; Bloom, J.D.; Thompson, A.S.; Winzenberg, K.N. J. Am. Chem Soc. 1982, 104, 4015. (c) Ko, S.S.; Klien, L.L.; Pfaff, K.-P.; Kishi, Y. Tetrahedron Lett. 1982, 23, 4415.